Complete the following table with the appropriate information using the periodic table.

| Element | Symbol | Atomic Number | \# of Protons | \# of <br> Neutrons | Mass Number (amu) | \# of <br> Electrons |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Hydrogen |  |  |  | 1 |  |  |
|  | He |  |  |  | 4 |  |
|  | Na |  |  | 12 |  |  |
| Potassium |  |  |  |  | 39 |  |
|  |  | 20 |  | 20 |  |  |
|  |  |  | 26 | 30 |  |  |
|  | Rb |  |  |  | 86 |  |
| Carbon |  |  |  | 6 |  |  |
| Xenon |  |  |  |  | 131 |  |
|  |  |  |  | 16 |  | 15 |
|  | At |  |  | 126 |  |  |
|  |  |  |  |  | 14 | 7 |
| Copper |  |  |  |  | 64 |  |
|  | Ne |  |  | 10 |  |  |
|  |  |  | 23 | 27 |  |  |

## Challenge:

What do you think would happen if you changed the number of electrons of a given atom?

What do you think would happen if you changed the number of protons of a given atom?

What do you think would happen if you changed the number of neutrons of a given atom?

